

14

ATKT/Additional examination  
B.N. Bandodkar College of Science, Thane  
Four<sup>th</sup> Semester End examination, ~~March~~-2015  
USCH402 June 2015  
June

Duration: 2 hrs and 30 min.

Total Marks:75

N. B. :

1. All the questions are compulsory
2. Figures to the right indicates full marks
3. Use of log table/ non programmable calculator is allowed

- Q.1 a Write a note on HSAB concept. 4
- OR
- a Explain conjugate acid-base pair. 4
- b Write a note on hydrogen bonding. 4
- OR
- b Write a note on physical and chemical properties of metal carbonyls. 4
- c Explain trends in the following properties 4
- i) Atomic size ii) Metallic character iii) Density iv) Conductivity  
with respect to group 15 elements.
- OR
- c Explain bonding in metal carbonyls on the basis of VBT. 4
- d Explain acid and base on the basis of Arrhenius theory and mention its advantages. 4
- OR
- d Explain limitations of Arrhenius theory of acid base. 4
- e Give the applications of organometallic compounds. 4
- OR
- e Write a note on 18 electron rule. 4
- Q.2 a Give any one preparation, properties, structure and bonding in  $XeF_6$  4
- OR
- a Write a short note on chlorophyll. 4
- b Give any method of preparation, properties and uses of thiocyanogen 4
- OR
- b Discuss the trends in chemical properties of 17 group elements . 4
- c Discuss the structure of  $XeOF_4$  4
- OR
- c Write a short note on myoglobin. 4
- d Give the electronic configuration of the elements of group 18. 4
- OR
- d Discuss the similarities between halogens and pseudohalogens. 4

- e Explain the Variation of colour among the halogens. 4
- OR**
- e Explain the Chlorine has higher electron affinity than fluorine. 4
- Q. 3 a Define the corrosion and explain different types of Corrosion? 4
- OR**
- a What is water pollution? Explain its causes. 4
- b Explain the electrochemical theory of corrosion with suitable example. 4
- OR**
- b Explain in detail the treatment of water waste. 4
- c What are the factor which influence the corrosion? 4
- OR**
- c What is soil pollution? explain its control measures. 4
- d Explain the cathodic and anodic protection of metal. 4
- OR**
- d Giving an example ,explain the role of an individual in overcoming crises of Pollution. 4
- e Write a short note on electroplating and organic coating method for protection of corrosion. 4
- OR**
- e Write a short note on Ganga action plane 1985. 4
- Q. 4 a Give the electron configuration of last three elements of group 16. 3
- OR**
- a What are organometallic compounds ?Give its two examples. 3
- b Give the preparation of XeF<sub>2</sub>. 3
- OR**
- b Give the properties of cynogen. 3
- c What is direct and indirect loss <sup>of corrosion,</sup> explains with suitable example? 3
- OR**
- c What are the control measures of water pollution? 3
- d Give two methods of preparations of Ni(CO)<sub>4</sub>. 2
- OR**
- d Give any two chemical reactions of Ni(CO)<sub>4</sub>. 2
- e Name the type of hybridization in XeF<sub>4</sub>. 2
- OR**
- e Give the different types of oxyfluoride. 2
- f What is differential aeration principle? 2
- OR**
- f Give the effects of soil pollution. 2

\*\*\*\*\*