

**POST GRADUATE DIPLOMA IN APPLIED ANALYTICAL CHEMISTRY**  
( 1<sup>st</sup> Sem. Examination 2006-2007)

Date: 22.12.2006

**PAPER- II**

Marks: 50

Note: Illustrate your answers with neat sketches, wherever necessary.

Q.1) Attempt any Three of the following:

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- Define Infra-red Spectroscopy. Describe the various molecular vibrations in the technique. What is the major requirement for infra-red absorption?
- Describe in brief nuclear magnetic resonance spectrophotometer. What information can be obtained from the NMR absorption peaks?
- Explain the principle and working of double beam spectrophotometer.
- Draw a schematic diagram of Gas Chromatography. Explain in brief the function of each component of G.C.
- Name the types of columns used in G.C. What do you mean by reverse phase and normal phase chromatography in HPLC?

Q.2) Attempt any Four of the following:

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- What is chemical shift? Why tetra methyl silane (TMS) is used as reference standard in NMR?
- What are the advantages and limitations of Gas Chromatography?
- Discuss the important applications of UV Visible Absorption Spectroscopy.
- Give the comparison between GC and HPLC.
- What is finger print region? Describe the shielding and de-shielding effects involved in NMR spectroscopy.

Q.3) Write note on any Three of the following:

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- Detectors used in HPLC
- Thermal conductivity detector
- Flame ionization detector
- Various detectors in UV visible spectrophotometer
- Role of calibration of instruments in UV visible spectrophotometer

Q.4) Write the position of peaks in IR and NMR spectroscopy for the following compounds:

**03**

- p-nitrophenol
- p-nitrobenzoic acid